

This article was downloaded by:

On: 22 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713454007>

Two new xanthenes from *Polygala crotalarioides*

Yan Hua^a; Chang-Xiang Chen^b; Yu-Qing Liu^b; Jun Zhou^b

^a Southwest Forestry College, Kunming, Yunnan, China ^b State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, The Chinese Academy of Sciences, Kunming, Yunnan, China

To cite this Article Hua, Yan , Chen, Chang-Xiang , Liu, Yu-Qing and Zhou, Jun(2007) 'Two new xanthenes from *Polygala crotalarioides*', Journal of Asian Natural Products Research, 9: 3, 273 – 275

To link to this Article: DOI: 10.1080/10286020600650040

URL: <http://dx.doi.org/10.1080/10286020600650040>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Two new xanthenes from *Polygala crotalarioides*

YAN HUA^{†,‡}, CHANG-XIANG CHEN[†], YU-QING LIU[†] and JUN ZHOU^{†*}

[†]State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, The Chinese Academy of Sciences, Kunming 650204, Yunnan, China

[‡]Southwest Forestry College, Kunming 650224, Yunnan, China

Two new xanthenes, 1,6,8-trihydroxy-7-methoxy-2,3-methylenedioxyxanthone (**1**) and 1,6-dihydroxy-7,8-dimethoxy-2,3-methylenedioxyxanthone (**2**), have been isolated from the roots of *Polygala crotalarioides*. Their structures were elucidated by spectral and chemical methods.

Keywords: *Polygala crotalarioides*; Polygalaceae; Xanthenes; bioactivities

1. Introduction

Polygala crotalarioides Buch. Ham. (Polygalaceae) is known as a tonic folk medicine in Yunnan Wa nationality [1]. Its bioactivities attracted us to investigate its chemical constituents. During our studies on this plant, two new xanthenes (**1** and **2**) were isolated from the roots of *Polygala crotalarioides*. The structural elucidation is reported in this paper (figure 1).

2. Results and discussion

Compound **1** was obtained as yellow needles, mp 250–252°C. The molecular formula was assigned as C₁₅H₁₀O₈ on the basis of HREIMS (*m/z* 318.0547). Its ¹³C NMR spectral data were indicative of a hexasubstituted xanthone, having one methylenedioxy moiety (δ 103.4, t, –O–CH₂–O–), one methoxyl and three hydroxyls. The carbonyl signal at δ 183.9 in the ¹³C NMR spectrum indicated a double-chelated carbonyl, meaning there were two hydroxyls attached at C-1 and C-8 [2]. The UV spectrum of **1** showed absorptions at 208, 255, 324 nm, which exhibited a bathochromic shift after addition of NaOAc, indicating the presence of a hydroxyl group at C-3 or C-6 [3]. The position of methoxy group in compound **1** was determined to be di-*ortho* substituted type based on the signal at δ 61.3 in the ¹³C NMR spectrum [2]. The δ value of C-2 at 130.1 ppm showed that the oxygenation pattern was 1, 2, 3 rather than 1, 2, 4 or 1, 3, 4 type [4]. The ¹H NMR spectrum of **1** showed two aromatic

*Corresponding author. Email: jzhou@mail.kib.ac.cn

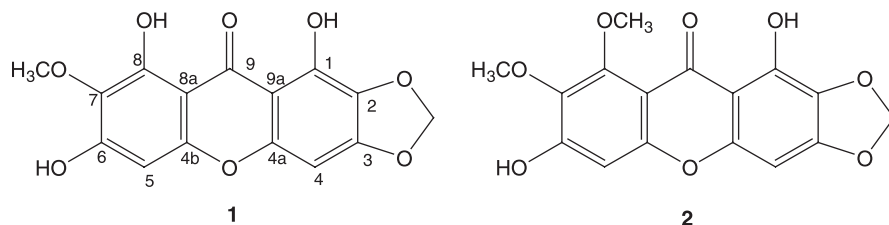


Figure 1. Structures of compounds **1** and **2**.

proton singlets at δ 6.59 and 6.74, assignable to H-4 and H-5 respectively. The three-proton singlet at δ 3.93 should be assigned to the methoxy group at C-7. On the basis of the above evidence, the structure of **1** was elucidated as 1,6,8-trihydroxy-7-methoxy-2,3-methylenedioxyxanthone.

Compound **2**, amorphous yellow powder, mp 275–277°C, was analyzed for $C_{16}H_{12}O_8$ by HREIM at m/z 332.0639 $[M]^+$. Its ^{13}C NMR spectral data were also indicative of a hexasubstituted xanthone, having one methylenedioxy moiety (δ 103.2, t, $-O-CH_2-O-$), two hydroxyls and two methoxyls. The signal at δ_C 181.2 indicated a free hydroxyl at C-1 or C-8, chelated with the carbonyl group [2]. As **1**, the NaOAc-induced shift was also observed in the UV spectrum of **2**, indicating the presence of a hydroxyl group at C-3 or C-6 [3]. The signals of two methoxyls in the ^{13}C NMR spectrum at δ 61.3 and 62.0 indicated that both were di-*ortho* substituted. In the ROESY spectrum, there was a correlation between the two methoxyl singlets at δ 3.90 and 4.12, requiring the presence of 7 and 8-substituted methoxy groups. The methylenedioxy group at C-2 and C-3 was also confirmed by the spectral comparison with that of **1**. The 1H NMR spectrum of **2** showed signals at δ 6.57 (1H, s) and 6.93 (1H, s), assignable to the protons at positions 4, 5, respectively. Hence, the structure of **2** was concluded to be 1,6-dihydroxy-7,8-dimethoxy-2,3-methylenedioxyxanthone.

3. Experimental

3.1 General experimental procedures

Melting point was measured on a Koffler melting point apparatus produced by Sichuan University (China) and is uncorrected. UV spectra were measured on a UV-210 spectrometer. EIMS and HR-EIMS were recorded on a VG Auto Spec-3000 mass spectrometer. NMR spectra were recorded in pyridine- d_5 , on a Bruker AM-400 or DRX-500 instrument at room temperature, using TMS as an internal standard.

3.2 Plant material

The roots of *Polygala crotalarioides* were collected from Yun County, Yunnan Province, China, in May 2002. The plant was identified by Prof Shu-Kun Chen at Kunming Institute of Botany, the Chinese Academy of Sciences, where a voucher specimen (No. 0154671) is deposited.

3.3 Extraction and isolation

The dried roots (1 kg) of *P. crotalarioides* were extracted with 75% EtOH four times under reflux. After removal of the solvent *in vacuo*, the aqueous solution was passed through a

HPD-100 column and the absorbed materials were eluted with water, 75% aqueous ethanol and ethanol, successively. The 75% ethanol eluate was concentrated *in vacuo* to give a residue (96 g), which was chromatographed on a silica gel (200–300 mesh) column with CHCl_3 –MeOH– H_2O (7:3:0.5, v/v/v) to afford ten fractions. Fraction 7 was chromatographed on Si gel with CHCl_3 –MeOH (100:1) and resubjected to Si gel column chromatography eluting with petroleum ether– Me_2CO (4.5–1, v/v) to afford compounds **1** (6 mg), **2** (65 mg).

3.3.1 6,8-trihydroxy-7-methoxy-2,3-methylenedioxyxanthone (1). Yellow needles, mp 250–252°C. UV (MeOH) λ_{max} nm: 208, 255, 323; + NaOAc: 209, 254, 351. EIMS (m/z): 318 $[\text{M}]^+$ (55), 303 (67), 288 (100), 275 (58), 149 (16). HREIMS: 318.0547 $[\text{M}]^+$ (calcd for $\text{C}_{15}\text{H}_{10}\text{O}_8$, 318.0591). ^1H NMR (400 MHz, $\text{C}_5\text{D}_5\text{N}$): δ 6.74 (1H, s, H-5), 6.59 (1H, s, H-4), 6.12 (2H, s, –O– CH_2 –O–), 3.93 (3H, s, OMe-7); ^{13}C NMR (100 MHz, $\text{C}_5\text{D}_5\text{N}$): δ 183.8 (s, C-9), 160.8 (s, C-3), 158.5 (s, C-6), 158.4 (s, C-8), 155.7 (s, C-4b), 153.6 (s, C-1), 151.5 (s, C-4a), 143.1 (s, C-7), 130.1 (s, C-2), 104.1 (s, C-8a), 103.4 (t, –O– CH_2 –O–), 101.6 (s, C-9a), 99.7 (d, C-5), 90.0 (d, C-4), 61.3 (q, OMe-7).

3.3.2 6-dihydroxy-7,8-dimethoxy-2,3-methylenedioxyxanthone (2). Amorphous yellow powder, mp 275–277°C. UV (MeOH) λ_{max} nm: 210, 249, 322; + NaOAc: 206, 247, 358. EIMS (m/z): 332 $[\text{M}]^+$ (55), 303 (67), 288 (100), 275 (58), 149 (16). HREIMS: 332.0639 $[\text{M}]^+$ (calcd for $\text{C}_{16}\text{H}_{12}\text{O}_8$, 332.0675). ^1H NMR (400 MHz, $\text{C}_5\text{D}_5\text{N}$): δ 6.93 (1H, s, H-5), 6.57 (1H, s, H-4), 6.10 (2H, s, –O– CH_2 –O–), 3.90 (3H, s, OMe-7), 4.12 (3H, s, OCH_3 -8); ^{13}C NMR (100 MHz, $\text{C}_5\text{D}_5\text{N}$): δ 181.2 (s, C-9), 159.7 (s, C-6), 159.5 (s, C-3), 155.0 (s, C-4b), 154.9 (s, C-8), 153.9 (s, C-1), 152.6 (s, C-4a), 140.4 (s, C-7), 129.7 (s, C-2), 108.2 (s, C-8a), 105.8 (s, C-9a), 103.2 (t, –O– CH_2 –O–), 100.5 (d, C-5), 88.9 (d, C-4), 61.3 (q, OMe-7), 62.0 (q, OMe-8).

Acknowledgements

The authors are grateful to the Analytical Group of the Laboratory of Phytochemistry, Kunming Institute of Botany, Chinese Academy of Sciences, for the spectral measurements.

References

- [1] B.X. Xiang, P.F. Zhang, Y.H. Xiang. *Gui Zhou Sci.*, **13**, 24 (1995).
- [2] I. Miura, K. Hostettmann, K. Nakanishi. *Nouv. J. Chim.*, **2**, 653 (1978).
- [3] A.A. Lins Mesquita, D. De Barros Correa, O.R. Gottlieb. *Anal. Chim. Acta*, **42**, 311 (1968).
- [4] T.R. Pinheiro, V.C. Filho, A.R.S. Santos. *Phytochemistry*, **48**, 725 (1998).